

Evaluation of Addition of Carbon Nano Materials Over Properties of an Elastomeric Matrix

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Abstract: In the present work, the effect of the addition of different types of carbon nano structures on the mechanical, thermomechanical and thermal properties of a radial structure of styrene-butadienestyrene (SBS_R) copolymer matrix is reported. Different carbon nanostructures were used as nano-reinforcements: expanded graphite (XG), graphene oxide (GO), reduced graphene oxide (RGO) and exfoliated graphene (EG). These carbon structures present various functional groups, such as carbonyl, epoxy, and others, which are the responsible for the interaction between the polymer matrix and the nano particles. The compatibility induced between the nanomaterials and the elastomeric matrix favors the stable dispersion of the nanocomposites during their obtention process. For instance, the addition of GO increased in 10 and 16% the tensile strength and storage modulus of the nanocomposites. The fracture surface patterns in the nanocomposites after the tensile test was observed by scanning electron microscopy. Also, the dynamic mechanical analysis (DMA) and thermal characterization showed differences in the viscoelastic behavior of the reinforced nanocomposites with different carbon nanomaterials.

Keywords: carbon nanomaterials, thermal properties, styrene-butadiene-styrene

1.Introduction

The development of high-performance polymer composites has become a challenge for materials researchers, who work on obtaining multifunctional materials for various purposes [1, 2]. Composite materials of polymeric matrices exhibit improvements in their mechanical, thermal and chemical properties when modified with reinforcement materials, due to the fact that their interfaces are reenriched by the synergistic effect between the matrix and the nanostructure; however a typical micro reinforcement usually creates better interfaces and, consequently, produces notable improvements in mechanical properties, as well as in chemical and environmental resistance [1-3]. On the other hand, some of the most important challenges on this matter, such as easing the nano-reinforcement dispersion or impregnation of fibers and nanoparticles, as well as dealing with the high volume of microreinforcement particles, have not been yet solved entirely [3]. The primary function of the polymeric matrix in a composite is to provide elasticity, flexibility, strength, as well as other unique physical properties. There is a wide diversity of polymeric matrices like elastomers, thermoplastics, and epoxies with different stereochemical configurations like linear, radial, and multi-radial block copolymers, hydro, and aerogels [1]. Since the typical poor reinforcement-matrix compatibility implies deficient mechanical properties, some strategies have been explored in order to overcome this problem on the composite preparation, such as chemical functionalization or using coupling agents [4].

Carbon allotropes are classified according to the type of their bonds, for instance in sp² hybridization, graphite, graphene and carbon nanotubes [2]. Graphene is recognized as the hardest material in the world, with a Young's modulus value of 1 TPa and a breaking strength of 130 GPa, according to several references [5-8].

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Graphene has excellent mechanical, thermal, and electrical properties. This carbon nanomaterial is one of the most used and preferable reinforcers, standing out over nano clays like montmorillonite sodium (Na-MMT), or layered double hydroxide (LDH), because of its high surface area, as well as its notable thermal and electrical conductivity; for these reasons, nowadays, graphene is considered as the best nano filler for polymeric matrices [9-11]. However, pristine graphene is not compatible with organic polymers and does not form homogeneous components [9, 11]. In order to overcome these shortcomings, carbon-based nanoreinforcers, such as carbon black, exfoliated graphite, carbon nanotubes (CNT), graphene nano sheets, and functionalized carbon nanotubes (FCNT) have been used for the preparation of polymer nanocomposites [1, 8 12, 13].

On the other hand, there are many studies on expanded and exfoliated graphite composites based on a vast range of polymers including epoxy, poly methyl-methacrylate (PMMA), polypropylene (PP), linear low-density polyethylene (LLDPE), high-density polyethylene (HDPE), polystyrene (PS), Nylon, polyaniline, phenylethynyl-terminated polyimide, and silicone rubber [1, 6, 14]. Additionally, several mechanisms have been proposed to incorporate the polymer into the core of multilayer materials [7]. For instance, the method of intercalation by in situ polymerization, where graphene or modified graphene is first swollen within the monomer's liquid, then a suitable initiator is diffused, and the polymerization is initiated by heat or by radiation. Besides, the solution intercalation method is based on the selection of a suitable solvent in which the polymer is solubilized, and the layers of graphene or modified graphene are swollen by the polymer chains. The main advantage of this method is that it allows the synthesis of intercalated nanocomposites based on polymers with low or even no polarity. Then, the polymer is adsorbed on the delaminated sheets, and when the solvent evaporates, the sheets are reassembled, placing the polymer in the sandwich to form the nanocomposites. In the melt intercalation method, no solvent is required, and graphite or modified graphene is mixed with the polymer matrix in a molten state. In this intercalation technique, the copolymer is mixed mechanically with graphite or modified graphene at elevated temperatures [1]. The graphene/polymer composite is mostly prepared using solvent mixing and in situ polymerization [7].

In this work, graphene derivatives were added into the elastomeric matrix through the molten mixing method, evaluating the effect of the chemical structure of the maxtrix (linear and radial) on the final properties. The effect of carbon nanoparticles on the mechanical, thermomechanical, and thermal properties of the elastomeric matrix were also studied.

2. Materials and methods

2.1. Materials

The carbon nanomaterials used in the present research were expanded graphite (XG), graphene oxide (GO), reduced graphene oxide (RGO), and exfoliated graphite (EG), which were obtained by synthesis methodologies previously published [3, 15-19]. The elastomeric matrix, a styrene-butadiene-styrene copolymer with radial structure (SBS_R) with 30 % of total styrene content and a viscosity of 28 cSt in Toluene, was provided by Dynasol Elastomeros, (Altamira, Tamaulipas, Mexico).

2.2. Composites preparation

The nanocomposites were prepared by adding each nanomaterial into the polymer matrix at 0.01 wt. % and 0.1 wt. %. Polymer and nanomaterials were melt-mixed by using a Brabender PL2000 torque rheometer (Plasti-corder). The mixing was carried out at 150°C and 50 rpm for 15 min. After that, the materials were compress-molded in a DAKE press, at 160°C and 14 kN for obtaining 3.17 mm-thick plates, from where samples for dynamic mechanical analysis and mechanical test were cut according to the description in subsequent sections.

2.3. Composites characterization

Scanning electron microscope (SEM)

The fracture micrographs of scanning electron microscopy (SEM) were obtained using HITACHI

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TM-1000 equipment (Hitachi, Tokyo, Japan) through backscattered electrons in a low vacuum at 15 kV. Samples with N₂ were mounted on metal stubs and vacuum-coated with gold using a sputter coater (DENTON VACUUM Desk V HP), with a voltage operation of 20 kV.

2.3.1. Mechanical testing

The stress-strain curves were obtained using dog-bone shaped specimens in an Instron 3384 universal machine, with a crosshead rate of 5 mm/min and a load cell of 10 kN, according to ASTM D638, with 5 replicas.

2.3.2. Thermogravimetric Analysis (TGA)

The thermal stability of the composites was evaluated by means a TA Instruments STD Q600 thermogravimetric analyzer under a nitrogen atmosphere with a flow of 100 mL/min, in a temperature range of 30 to 700°C. The heating rate was 10°C/min, and the sample weight was 10 mg approximately.

2.3.3. Dynamic mechanical tests (DMA)

The dynamic mechanical analysis (DMA) was carried out in a TA-Instruments model Q800 using a dual cantilever clamp, in multifrequency mode with an amplitude of 20 mm and frequency 1 Hz, over the temperature range of -85 to 150°C. The heating rate was 5°C/min. The sample size was 20*14*3 mm³, and it was carried out in duplicate (the geometry was based on ASTM D4065).

2.3.4 X-ray diffractometer (XRD)

For X-ray diffraction analysis, thin films of nanocomposites were prepared, and their diffraction patterns were collected between 1 to 40°, using step- scanning mode of 0.016 grades (20) and 60 s counting time at each step with a Lynxeye detector on PANalytical XPert'PRO diffractometer equipped with an X'Celerator detector and $CuK\alpha$ radiation.

3. Results and discussions

3.1. Mechanical properties

The results obtained from the uniaxial tensile testing of the composites samples are shown in Table 1. As can be observed, the tensile strength (TS) of SBS_R matrix and nanocomposites decrease compared with SBS_R pristine.

The addition of 2D nano materials generated changes in the tensile strength (TS) of the nanocomposites SBS_R-GO 0.01 wt. %, and SBS_R-XG 0.01 wt. %, which exhibited increments of 16 % and 2 % in this property, respectively. The 2D carbon nanomaterials used as reinforcements have similar 2D structures to the pristine graphene, which outstanding properties are attributed to the high stability of the lattice built by the sp² bonds [20]. The functional groups created during the synthesis, for example, in GO and XG, disrupt the sp² lattice and, consequently, the mechanical properties are impacted. In the nanocomposites, the remainder spy lattice present in GO is enough to use it as reinforcement, and it is responsible for improving the mechanical properties. Also, the functional groups present in these materials (OH, COOH, C=O) offer an advantage in terms of compatibility and dispersion compared to that of the pristine graphene, whose poor capability to form stable solutions prevents the formation of stable and homogenous dispersions and, therefore, restricts its application in the field of nanocomposite technology [4, 11, 21].

Tan et al. reported the improvement of the mechanical properties of SBS nanocomposites by GO. They reported up to 24.2 % improvements on the TS in the SBS_R polymeric matrix reinforced with 3 wt. % of functionalized graphene (f-G) [5]. According to those results, f-G improved the TS and the elongation at break. In our case, SBS_R-GO 0.01 wt. % and SBS_R-XG 0.01 wt. % match that behavior, evidencing an increase in both parameters.

As can be inferred from the results of the nanocomposites reinforced with 0.1%, increasing the reinforcer load sacrifices the stiffness of the composites, but results in a tougher and more elastic



behavior, as shown in Table 1.

Table 1. Mechanical properties of the SBS_R, and 2D nanocomposites

Sample	wt. %	Modulus at 300% MPa	Std deviation, MPa	Strain %	Std deviation, %	Tensile strength MPa	Std deviation, MPa
SBS _R	0	5.27	0.23	348.48	11.9	5.58	0.43
SBS _R -GO	0.01	3.09	0.21	498.19	18.7	6.1	0.46
	0.1	5.22	0.22	356.07	12.6	5.36	0.41
SBS _R -RGO	0.01	4.34	0.24	319.80	11.7	4.9	0.32
	0.1	5.28	0.29	316.37	10.8	5.2	0.39
SBS _R -EG	0.01	3.32	0.19	416.6	13.0	4.27	0.36
	0.1			255.01	9.45	3.83	0.27
SBS _R -XG	0.01	3.09	0.17	583.72	20.7	5.71	0.41
	0.1	3.83	0.17	451.87	17.3	5.05	0.38

3.2. Fracture characterization

Figure 1 presents the SEM micrographs of the fractured surfaces of the elastomeric matrix, as well as those of the nanocomposites. Figure 1-a shows the fracture surface of SBS_R elastomer. The addition of modified graphene particles in an elastomeric matrix present micrometrical dimension and a flake form that did not show any exfoliation [12].

The Figures 1b-e show changes compared with SBS_R elastomer in the fracture surface after the addition of 2D nanocarbon materials. Mainly, the SBS_R-GO 0.01 wt. % nanocomposite showed the marked, notable patterns related to tearing lines, which are characteristic in fracture of elastomers. These changes are related to the response to the stress transfer from the matrix to the reinforcer [22, 23].

The geometry in 2D carbon nanomaterials plays a crucial role during the stress transfer [20, 24]. When a load is applied to the nanocomposite, the polymer matrix begins to deform around the graphenederivative sheet. The deformation origins a shear force between the polymer and the 2D nanomaterials concentrated in the vicinity of the sheet. At the point where the strain of the 2D nanomaterials and the polymer reach the same shear strain, a reinforcement effect comes from the differences in elastic modulus among the polymer and the 2D carbon nano-materials. When both are at the same strain, the stress in 2D sheets is more significant than in the matrix [20]. Also, the distortions and the resultant defects of GO result in a rough and winkled topology on the nanoscale which ensures a strong elastomerto-filler interaction [25]. Finally, the fracture surface of the nanocomposites with the best response in TS (SBS_R-GO 0.01 wt. % and SBS_R-XG 0.01 wt. %) evidence not obvious agglomerated 2D graphene derivatives, which points towards the best combination between dispersion and transfer of mechanical properties from nanomaterials to elastomer. Moreover, the functional groups present in these nanomaterials points that the high state of oxidation in the GO, and the respective oxidation in the case of the XG, impacts positively on the TS. Consequently, the river-like fracture patterns exhibited on these composites confirmed what was mentioned by other authors [26, 27]. Similar to our study, a previous work reported agglomerate formation during melt-compounding when carbonaceous filler was added to SBS_R triblock matrix, finding also that the amount of filler plays an important role to define its level of dispersion over the elastomeric matrix [28].



According to some references [29-31], the functional groups in the carbon reinforcer nanoparticles induce physical interactions between the elastomeric matrix and the nanoreinforcer; these matrix-nanoparticles interactions improve the dispersion. In the other hand, chemical interactions between the carbon nanoparticles and the SBS matrix may be triggered when melt mixing process is employed [32].

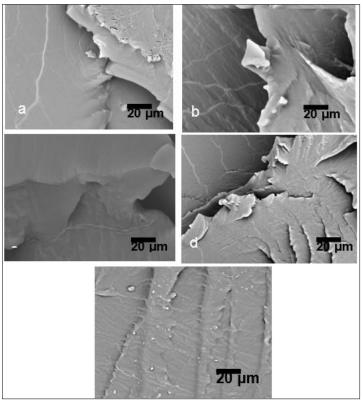


Figure 1. Fracture surface micrographs of a) SBS_R, of SBS_R nanocomposites reinforced with b) GO 0.01 wt. %; c) RGO 0.01 wt. %; d) EG 0.01 wt. %; e) XG 0.01 wt. %

3.3. Thermal analysis of the SBS_R and its nano-composites

The thermal behavior of the nanocomposites was evaluated in order to analyze the influence of the addition of different nanomaterials into the elastomeric matrix. All composites exhibit a single step thermal decomposition from 350 to 450°C, attributed to the decomposition of the elastomeric matrix backbone [28]. Since all the composites behave similarly, making hard to identify significative changes, the first derivative of the loss weight was plotted. Figure 2 (left)-(right) depicts the DTG curves for the nanocomposites and SBS, from where it is observed that the degradation temperature of the composites was higher than that of the neat SBS_R. This observation suggests that an interaction is established between the elastomeric matrix and the carbon nanoparticles at the interface, which would decrease the mobility of the polymer chains near the interface and increase the thermal stability of the composites. Therefore, different thermal behavior resulted in the composites as different nanomaterials were added to the elastomeric matrix. Although the total content of reinforcer was fixed, the thermal stability decreased in the order GO>XG>RGO>EG>SBS_R [5], at 0.01% wt in neat SBS_R, with a slight increase on the thermal stability (4 °C) of the elastomeric matrix when GO was added to SBS_R.

Figure 2 right shows the enlargement of the DTG curves; the change in the thermal stability of the nanocomposites maybe caused by the fact that some amounts of reinforcer act as heat source, promoting thermal degradation. Thus, a real contribution of reinforcement into the matrix is reached.

Yuanquin et al. [6] reported the maximum thermal degradation temperature of RGO/HO-SBS_R composites, finding that it becomes slightly lower as the RGO content increases. Other report indicates



that the heating rate has a significative effect on the thermal stability of SBS_R -carbon nanotubes composites, having a negative effect which decreased the elastomeric thermal stability [33].

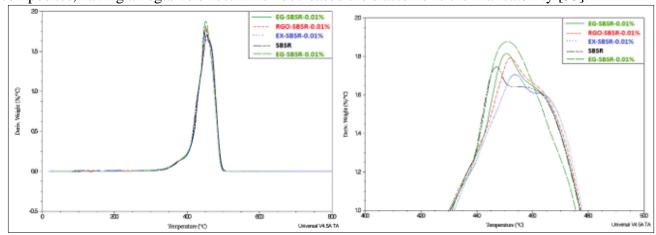


Figure 2. DTG curves of SBS_R and composites at 0.01% (left), Enlargement of the DTG curves of SBS and composites at 0.01% (right)

3.4. Dynamic Mechanical Analysis (DMA)

The storage modulus (E') and Tan δ curves as a function of the temperature of the composites were obtained by DMA. The evaluation of these properties will be focused on the influence of adding different reinforcers into the elastomeric matrix of radial structure. By using the storage modulus curve, the glass transition temperature (Tg) and the crosslinking density (V_e) were determined and reported in Table 2. However, the analysis of the curves will only focus on the reinforced SBS_R composites that showed the best storage modulus behavior.

As observed in Table 2, the crosslinking densities (V_e) of all the SBS_R materials suggest better matrix-reinforcer interactions as a result of having polymeric chains of star structure, which is consistent with previous studies regarding radial structure matrices [34]. Moreover, the presence of oxygenated functional groups does not interfere with the vulcanization process of the elastomeric matrix. Conversely, when GO was added, the vulcanization process was favored, as shown by V_e , followed by RGO>XG>EG in the case of SBS_R. These results can be related to the mechanical properties; first, as other authors reported [34, 35], the presence of the functional groups favor the dispersion. Second, the vulcanization is less affected were dispersion is appropriately acquired. For the SBS_R, it is possible to infer that the amount of reinforcer added was excessive, which impacted negatively on the vulcanization process. As the table 2 shows for the low concentration of 2D carbon in nanocomposites, the V_e is almost similar to that of the elastomeric matrix. In contrast, for the high percentage of reinforcer, the V_e falls considerably in all cases. However, these results need to be studied more deeply, as Stainer et al. reported [36], who found that the best performance of GO-natural rubber composites was achieved when a high amount (1 wt. %) of 2D carbon nanomaterial was added.

Table 2. Storage modulus, Tg, and crosslinking density of the SBS_R, and 2D nanocomposites

Material	Wt %	Storage modulus, (MPa)	Tg, °C	Crosslinking density, (Ve)
SBS_R	0	568	-80	2451.59
SBS _R -GO	0.01	644	-81	3028.43
	0.1	551	-79	3028.43
SBS _R -RGO	0.01	408	-80	2884.22
	0.1	372	-80	2451.59



SBS _R -EG	0.01	551	-82	1874.74
	0.1	684	-79	2163.17
SBS _R -XG	0.01	1032	-77	2018.96
	0.1	379	-83	2018.96

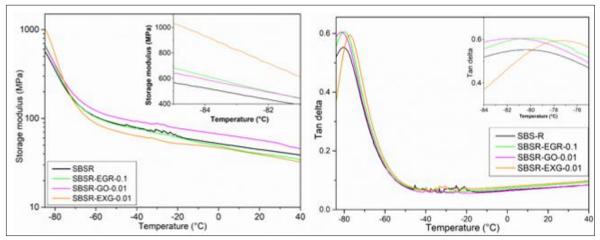


Figure 3. Storage modulus (E')(left) and Tan δ (right) for SBS_R and 2D nanocomposites

Figure 3-left shows the curves of the storage module of the nanocomposites that reached higher values of E' than the SBS_R polymer matrix. In the curves of the storage modulus, the highest value of E' (-85°C) correspond to the composites reinforced with EG, GO and XG. 82 % of increase with respect to SBS_R (E' = 568 MPa) is reached using only 0.01 wt% of XG (SBS_R-XG-0.01). On the other hand, increases of 13% and 20% were achieved with 0.01 wt% and 0.1 wt% of GO and EG, respectively. With respect to the increase of 13% due to the incorporation of GO, it produces great stiffness at low temperatures of the elastomer, as previously reported in a study where graphene oxide was also used to reinforce a natural rubber, and the same behavior was observed by adding this type of filler [37].

The Tan δ is determined by the ratio of loss (E'') and storage (E') modulus as a function of temperature [27]. This ratio reflects the reinforcement/matrix adhesion level. In the DMA, the maximum of the tan δ is related to Tg. The term Tg is the reversible transition in amorphous materials (including amorphous regions within semi-crystalline materials) from a molten or rubber-like state into a hard and relatively brittle, glassy state [38]. Table 2 presents the values obtained for Tg, and Figure 3 (right) shows the tan δ curves for the analyzed materials.

Wu et al reported that the addition of a carbonaceous filler to SBS_R elastomer did not improve the storage modulus in the glassy region, but in rubbery region it had the inverse effect. This phenomenon was attributed to the intercalation of the filler in the elastomeric matrix, which affected the mobility of the polymer chains [37].

3.5. Wide Angle X-ray Scattering

Composites with SBS_R polymer display a well-resolved diffraction peak superimposed on an amorphous background, as seen in Figure 4. The peak of maximum intensity at 2θ ~19.4° is attributed to the characteristic peak of pure graphite. The diffraction patterns were similar for all SBS_R-graphene composites, differing only in their intensities. Small decreases probably reflect reductions in remaining intercalants. Intercalation of the thermally exfoliated graphene (EG) is expected to increase the distance between graphene sheets [1, 2, 6]. The mixing temperature plays an important role affecting the interlayer spacing in the elastomeric matrix, showing that delamitation is induced at lower temperatures [12].



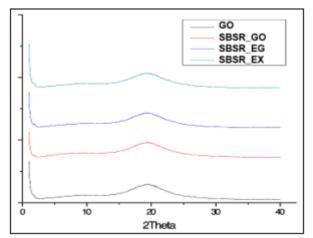


Figure 4. XRD spectra of SBS_R and SBS_R/graphene composites

4. Conclusions

The addition of different carbon nanoreinforcers over SBS_R elastomeric matrix has a variable effect on properties; GO y XG cause an increment on tensile strength (TS) attributed to a high stability of lattice built by the sp^2 bonds. In general mechanical properties show an increase compared with SBS_R pristine and an increasing of reinforcer load causes decrease of the stiffness of the composites. In the other hand, he addition of modified graphene particles to the nanocomposites exhibited a flake-shaped surface, and some evidence of exfoliation was found by micrographs of the fractured surface, due to the stress transfer from the elastomeric matrix to the reinforcer. Also, the geometry of carbon nanoparticles plays an important role in the stress transfer.

The thermal stability of nanocomposites was higher compared with SBS_R pristine, in a decreasing order GO>XG>RGO>EG. The DMA helps to determine the crosslinking density of nanocomposites, obtaining that prepared nanocomposites have a good matrix-nanoreinforcer interaction attributed to lake star structure of SBS_R . The storage modulus of nanocomposites was higher compared with elastomeric matrix and Tg value determinate using $Tan\ \delta$ curve shows that there is not a significative change in this property of nanocomposites compared with pristine elastomeric matrix.

Acknowledgments: The present project financial support from Tecnológico Nacional de México, with code 6446.18-P and MAD-PYR-2019-7433. The authors are grateful to to Ph.D. Marina Vega from the CGEO-UNAM for technical support in SEM analysis, also appreciate the support of undergraduate students, Karem María Moreno Rocha, Jesus Maya Perales, Luis Isaac Naranjo Hernández, and Diego Armando Payón.

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Manuscript received: 27.05.2020